

Jian-Zhong Cui,* Hong Zhang,
Dan Zhang, Hai-Tao Wang and
Hong-Ling Gao

Department of Chemistry, Tianjin University,
Tianjin 300072, People's Republic of China

Correspondence e-mail:
cuijianzhong@tju.edu.cn

Key indicators

Single-crystal X-ray study
 $T = 294$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
Disorder in main residue
 R factor = 0.049
 wR factor = 0.155
Data-to-parameter ratio = 13.0

For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

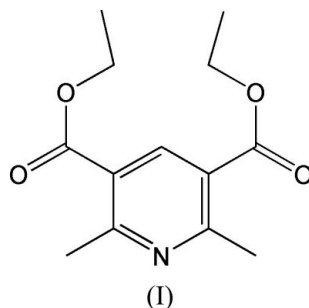
Diethyl 2,6-dimethylpyridine-3,5-dicarboxylate

The molecular structure of the title compound, $\text{C}_{13}\text{H}_{17}\text{NO}_4$, has no symmetry plane because one C atom of the ethyl group lies out of the plane defined by the other non-H atoms. In the crystal structure, the molecules stack along the a axis. Symmetry-related molecules are linked by a $\text{C}-\text{H}\cdots\text{O}$ hydrogen bond, forming zigzag chains extending in the b -axis direction.

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Comment

Dihydropyridine compounds are calcium ion channel blockers and the use of these compounds is generally beneficial (Böcker *et al.*, 1986). Aromatization of Hantzsch 1,4-dihydropyridines (1,4-DHP) has attracted considerable attention in recent years, essentially since the discovery that the metabolism of these drugs involves an oxidation step (Eynde *et al.*, 1995). Here we describe the synthesis (Lu *et al.*, 2001) and crystal structure of the title compound, (I).



The molecular structure of the compound (I) is shown in Fig. 1 and selected geometric parameters are given in Table 1. All of the non-H atoms in (I) lie almost in the same plane, except atom C10 which is out of the plane with a $\text{C}8-\text{O}2-\text{C}9-\text{C}10$ torsion angle of $123.2(17)^\circ$. This is probably because of the disorder of the ethyl group, and for this reason the molecule has no symmetry plane.

In the crystal structure of (I), the molecules stack along the a axis and are linked by $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds, forming polymer chains extending in the b -axis direction (Table 2, Fig. 2).

Experimental

The title compound, (I) was prepared according to the literature procedure of Lu *et al.* (2001). A mixture of diethyl 2,6-dimethyl-1,4-dihydropyridine-3,5-dicarboxylate, ferric chloride hexahydrate, acetic acid and water was heated at reflux for 1 h. After neutralization with an aqueous solution of sodium bicarbonate and extraction with chloroform, the title compound was obtained (yield 52%; m.p. 343–

344 K). Single crystals, suitable for X-ray diffraction analysis, were obtained by slow evaporation of an ethanol solution. IR (KBr, ν cm^{-1}): 2979, 2932, 1721, 1591, 1442, 1367, 1296, 1223, 1120, 1043, 771, 698.

Crystal data

$\text{C}_{13}\text{H}_{17}\text{NO}_4$
 $M_r = 251.28$
 Monoclinic, $P2_1/c$
 $a = 4.593$ (2) Å
 $b = 15.950$ (9) Å
 $c = 18.795$ (10) Å
 $\beta = 90.656$ (9)°
 $V = 1376.9$ (13) Å³
 $Z = 4$

$D_x = 1.212$ Mg m^{-3}
 Mo $K\alpha$ radiation
 Cell parameters from 1328 reflections
 $\theta = 2.5$ – 23.6 °
 $\mu = 0.09$ mm^{-1}
 $T = 294$ (2) K
 Block, colorless
 $0.24 \times 0.20 \times 0.16$ mm

Data collection

Bruker SMART CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.960$, $T_{\max} = 0.986$
 6854 measured reflections

2427 independent reflections
 1158 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.046$
 $\theta_{\text{max}} = 25.0$ °
 $h = -4 \rightarrow 5$
 $k = -18 \rightarrow 12$
 $l = -22 \rightarrow 22$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.049$
 $wR(F^2) = 0.155$
 $S = 1.00$
 2427 reflections
 187 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0658P)^2 + 0.2484P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.002$
 $\Delta\rho_{\text{max}} = 0.20$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.14$ e Å⁻³

Table 1

Selected torsion angles (°).

C8–O2–C9–C10	123.2 (17)	C11–O4–C12–C13	178.7 (3)
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Table 2

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C12–H12B \cdots O1 ⁱ	0.97	2.52	3.346 (4)	143

Symmetry code: (i) $-x + 2, y - \frac{1}{2}, -z + \frac{1}{2}$.

All H atoms were positioned geometrically and refined as riding atoms with C–H distances = 0.93–0.97 Å. For the aromatic and CH₂ H atoms $U_{\text{iso}}(\text{H}) = 1.2U_{\text{e}}(\text{C})$, and for the CH₃ H atoms $U_{\text{iso}}(\text{H}) = 1.5U_{\text{e}}(\text{C})$. The ethyl group was found to be disordered over two orientations, with occupancies of 0.39 (2) and 0.61 (2).

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

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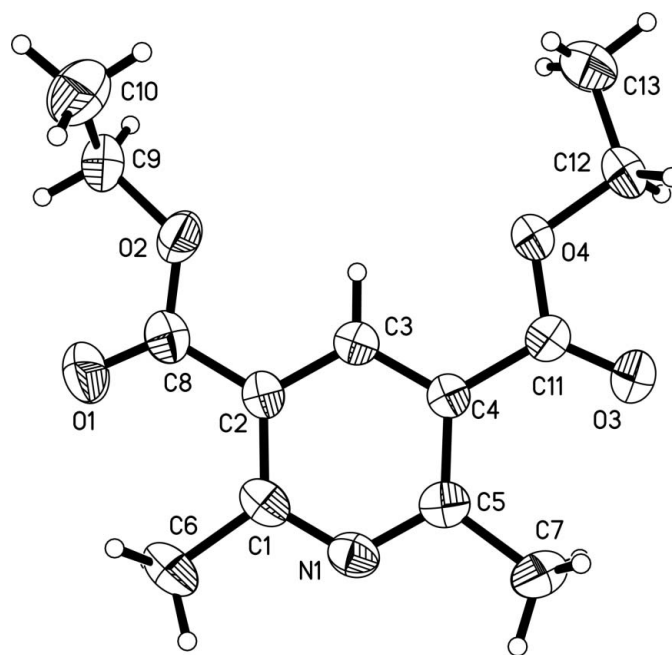


Figure 1

The molecular structure of compound (I), showing the atom-labeling scheme, with displacement ellipsoids drawn at the 35% probability level. Only the major component of the disordered ethyl group is shown.

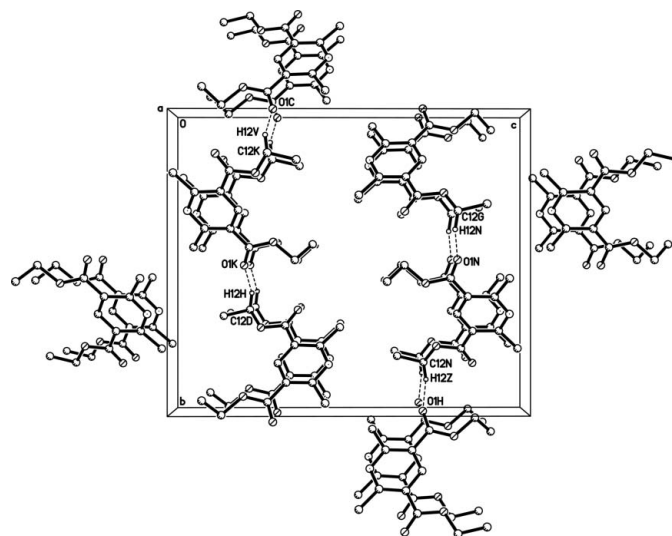


Figure 2

The crystal packing of compound (I), viewed along the a axis. Only the major component of the disordered ethyl group is shown. The C–H \cdots O hydrogen bonds are shown as dashed lines.

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